NEW FURANOSESQUITERPENES FROM EUMORPHIA SPECIES*

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Key Word Index—Eumorphia sericea; E. prostata; Gymnopentzia bifurcata; Compositae; new furanosesquiterpenes; new triterpene.

Abstract-Eumorphia sericea and E. prostata contained several new furanosesquiterpenes and a new triterpene aldehyde. Structures were elucidated mainly by spectroscopic methods and by some chemical reactions. Structures of three compounds however could not be established with certainty and are partially based on biogenetic considerations. Gymnopentzia bifurcata also contained known furanosesquiterpenes. The number of South African genera from the tribe Anthemideae, where these furanosesquiterpenes are characteristic, has now increased to eight. The importance of these observations for the botanical position of these genera is still not clear.

INTRODUCTION

The South African genus Eumorphia belongs to a group of genera in the tribe Anthemideae, whose position in the tribe is problematic [1]. Only E. dregeana DC has been investigated chemically [2]; this species contains no acetylenes, but characteristic furanosesquiterpenes (2-4) which are also present in Lasiospermum, Ursinia, Athanasia, Phymaspermum, Stilpnophytum and Peyrousia species [3-9]. We have now investigated two further Eumorphia species and Gymnopentzia bifurcata, which also belongs to the woody South African Anthemideae.

RESULTS AND DISCUSSION

The roots of Eumorphia sericea contained the furanosesquiterpenes 1 and 2 [8], while the aerial parts

Table 1. ¹H-NMR data of compounds 8, 9, 11 and 12 (δ -values, TMS as internal standard, 270 MHz, CDCl₃)

	8	9	11	12
1-H	1.25 s	1.27 s		
3-H	3.55 s	3.88 s		_
5-H	2.85 d	2.55 dd	2.40 dd	2.36 dd
5'-H	2.73 d	2.40 dd	2.25 dd	2.21 dd
6-H		2.15 ddddq	2.05 ddddq	2.05 ddddq
7-H	2.22 m	1.76, 1.63 m	1.81, 1.65 m	1.78, 1.63 m
8-H	1.82 m	2.77 m	2.78 m	2.77 m
9-H	4.93 dd		_	
11-H	$6.41 \ s(br)$	6.77 dd	6.77 dd	6.77 dd
12-H	7.38 dd	7.45 dd	7.45 dd	7.44 dd
13-H	7.41 s(br)	8.05 dd	8.04 dd	8.03 dd
14-H	1.34 s	0.98 d	1.04 d	1.01 d
15-H	1.35 s	1.44 s	_	~

^{11,12 = 1.6}; 11,13 = 0.7; 12,13 = 1.6.

Table 2. ¹H-NMR data of compound 28

	C_6D_6	+ Eu(fod) ₃	CDCl ₃
1-H)	4.07	2.55 m	
1'-H	1.87 m	2.73 m	
2-H ĵ		3.58 dddd	2.38 dddd
2'-H (2.14 m	3.74 dddd	2.16 dddd
3-H	9.43 t	11.64 s(or)	9.69 t
5-H	$5.21 \ d(br)$	5 64 d(hr)	$5.16 \ d(br)$
6-H	2.94 ddd	3.40 ddd	2.90 ddd
21 -H	$5.28 \ t(br)$	5.30 t(br)	5.10 t(br)
23-H	1.63 s(br)	$1.64 \ s(br)$	$1.62 \ s(br)$
24-H	1.72 s(br)	$1.73 \ s(br)$	$1.68 \ s(br)$
25-H	$1.00 \ d$	1.03 d	0.91 d
26-H	0.79 s	0.88 s	$0.800 \ s$
27-H	0.87 s	1.09 s	0.84 s
28-H	0.87 s	1.14 s	0.90 s
29-H	$1.54 \ s(br)$	1.84 s(br)	$1.60 \ s(br)$
30-H	$1.63 \ s(br)$	1.84 s(br)	1.68 s(br)

 $J(H_7)$: 1,2 = 7.5; 1,2' = 6; 2,2' = 16; 2,3 = 1.5; 5,6 = 10; 6,7 = 7; 18,25 = 7; 29,21 = 7.

afforded the hydrocarbons 19, 21 and 22, the furans 1-4 [3, 8], 6 [10], 7 and 10 [9] together with the aromatic compounds 23, 24 and two further furanosesquiterpenes both having the molecular formula C₁₅H₂₀O₄. While the more polar substance was a β -ketofuran (IR: 1690, 1570, 880 cm⁻¹; ¹H NMR: δ 8.05 dd, 7.45 dd and 6.77 dd),

Table 3. 1H-NMR data of compound 5

1-H	1.98 d	8-H	2.73 m	
3-H	$5.88 \ s(br)$	11-H	6.76 dd	
5-H	2.57 d	12-H	7.43 dd	
5'-H	2.46 d	13-H	8.00 dd	
6-H	1.87 ttq	14-H	0.95 d	
7-H	1.75 m	15-H	7.06 dd	

J (Hz): 5.5' = 17; 5.6 = 5.5; 5'.6 = 7.5; 6.7 = 7; 6.14 = 7; J (Hz): 1.15 = 3.15 = 1; 5.5' = 15; 5.6 = 6.5; 5.6' = 6.7 = 16,14 = 7; 11,12 = 1.5; 11,13 = 12,13 = 1.

^{*} Part 137 in the series: "Naturally Occurring Terpene Derivatives": for part 136 see: Bohlmann, F., Suwita, A. and Mabry, T. J. (1978) Phytochemistry 17, 763.

the less polar one was closely related to 6 as most of the NMR signals were very similar in both compounds (Table 1). However, instead of the two methyl doublets, two singlets were observed. The same was true for the more polar compound, its NMR spectrum was similar to that of 7. Singlets at $\delta 3.38$ and 3.55 respectively indicated an epoxide. Therefore the structures of the two furanes were most probably 8 and 9. By acid hydrolysis and periodate cleavage 9 was transformed into the acid 11, which established this assumption. We have named compound 9 eumorphinone and compound 8 cycloeumorphinone. A small amount of a triterpene aldehyde was also isolated, both from the roots and the aerial parts, with the molecular formula C30H50O. The spectroscopic data (Table 2) and biogenetic considerations are in agreement with structure 28. A possible

pathway to 28 could be the rearrangement of 25 (Scheme 1). However, the amount of material was not sufficient to establish the proposed structure.

The aerial parts of Eumorphia prostata contain germacrene C (20), 7, 13 [8, 9] and a further ketone with the molecular formula $C_{15}H_{18}O_3$. The NMR spectrum (Table 3) was in agreement only with structure 5. The polar fractions afforded a hydroxy ketone as shown by the IR spectrum. In the MS no molecular ion was detectable. The observed fragments however were in agreement with the molecular formula $C_{15}H_{20}O_3$. Intensive NMR studies finally led to structure 16 (Table 4). Decoupling experiments and shift reagents clearly showed that the compound was a trisubstituted cyclohexene derivative. The two protons at C-2 showed allylic coupling with 4-H and a vicinal one with 1-H, which is the coupling partner of the secondary methyl group. Furthermore 4-H showed a small coupling with 5-H. Dreiding-models show that the angle is nearly 90° ,

^{*} The stereochemistry of 5, 7, 8-10, 13 and 16-18 has not been elucidated.

Table 4. ¹H-NMR data of compound 16

	C_6D_6	+ Eu(fod) ₃	CDCl ₃
1-H	1.77m	2.14 ddddg	1.63 m
2α-H	2.98 dd(br)	$3.50 \ dd(br)$	2.98 m
2β-H	2.04 dd(br)	2.58 dd(br)	2.05 m
4-H	5.29 ddd	$6.48 \ s(br)$	5.27 ddd
5-H	3.64 dd(br)	4.50 dd(br)	3.92 dd(br)
6-H	1.77 m	2.30 m	2.05 m 1.63 m
7-H	0.98 d	1.07 d	1.13 d
9-H	1.18 s	1.78 s	1.31 s
10-H	1.13 s	1.72 s	1.29 s
13-H	6.72 dd	$7.05 \ s(br)$	6.79 dd
14-H	6.84 dd	$7.09 \ s(br)$	7.46 dd
15-H	$7.51 \ s(br)$	$8.03 \ s(br)$	$8.03 \ s(br)$

J(Hz): $1,2\alpha = 6$; $1,2\beta = 10$; 1,7 = 7; $2\alpha,2\beta = 16$; $2\alpha.4 = 2\beta,4 = 4,5 = 1.5$; 5,6 = 7.5; 13,14 = 1.5; 13,15 = 0.7; 14.15 = 1.

which would be in agreement with the coupling observed. The two remaining substituents can only be placed as shown in 16, as the double bond was not conjugated with the carbonyl group. The abnormal carbon skeleton of 16 may be formed by rearrangement of a precursor such as 14 (Scheme 2). We have named compound 16 eumorphistonol. Two isomeric ketones, which may be related to the intermediate 15 were isolated in minute quantities. All spectroscopic data showed that they might possess structures 17 and 18. However, the amount of material was too small to establish these structures. We therefore have started a synthesis of 17.

The roots of Gymnopentzia bifurcata, which also belongs to the endemic South African genera of the tribe Anthemideae, again contained 1 and 2 together with the dihydroflavone 30, while the aerial parts afforded 23, the flavones 29 and 30 [11, 12] as well as ngaione (6).

Summarizing the results it is of chemotaxonomical

Scheme 1.

Scheme 2.

Table 5. 1H-NMR data of compounds 17 and 18

	17	18
2-H	5.53 d(br)	2.31 m
4-H	2.92 m	2.90 m
6-H	$5.69 \ s(br)$	$5.67 \ s(br)$
9-H	$1.73 \ s(br)$	$4.20 \ s(br)$ $4.77 \ s(br)$
10-H	$1.78 \ s(br)$	$1.87 \ s(br)$
13-H	$6.81 \ s(br)$	$6.80 \ s(br)$
14-H	7.44 dd	7.46 dd
15-H	8.06 s(br)	$8.03 \ s(br)$

J(Hz): 2,3 = 9; 13,14 = 14,15 = 1.5.

interest that so many South African genera are characterized by the occurrence of furanosesquiterpenes. Morphological investigations are desirable to find out whether this is in agreement with anatomical features.

EXPERIMENTAL

IR spectra were measured in CCl₄; ¹H NMR: all assignments were established by decoupling experiments. MS were determined at 70 eV; optical rotations were measured in CHCl₃. The airdried plant material, collected in Natal, was extracted with Et₂O—petrol (1:2) and the resulting extracts were first separated by CC (Si gel, act. grade II) and further by TLC (Si gel, GF 254). Known compounds were identified by comparison of their IR and NMR spectra with those of authentic material.

Eumorphia sericea Wood et Evans (voucher 77/282). Roots (90 g) afforded 120 mg 1, 60 mg 2 and 5 mg 28 (Et₂O-petrol, 1:20): 240 g aerial parts yielded 10 mg 19, 10 mg 22, 10 mg 28, 5 mg 21, 10 mg 3, 15 mg 4, 15 mg 10, 100 mg 6, 100 mg 7, 400 mg 1 and 2 (ca 2:1), 90 mg 8 (Et₂O-petrol, 1:1), 200 mg 9 (Et₂O-petrol, 1:1), 5 mg 23 and 35 mg 24.

Cycloeumorphinone (8). Colourless oil, IR $v_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: > C = O

1720; furan 1570, 880. MS m/e (rel. int.): 264, 136 (M⁺, 2) (calc. for $C_{15}H_{20}O_4$ 264.136); 249 (M⁺—·CH₃, 2): 246 (M⁺

Eumorphinone (9). Colourless crystals mp 69.5° (Et₂O-petrol), IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: C=O 1720; furan ketone 1690, 1570, 1520, 880. MS) m/e (rel. int.): 264. 136 (M⁺, 0.1) (calc. for C₁₅H₂₀O₄ 264. 136); 249(M⁺—CH₃, 2): 246 (M⁺—H₂O, 0.4); 231 (246

100). 10 mg 9 in 2 ml dioxane were heated with 0.5 ml 2N $\rm H_2SO_4$ for 5 min. at 70°. After addition of 30 mg $\rm HIO_4$ the mixture was allowed to stand for 1 hr. The $\rm Et_2O$ extract was shaken with NaHCO₃ soln and the acid part was purified by TLC (Et₂O-petrol, 1:3) after esterification with CH₂N₂ 3 mg 12 was obtained, colourless oil, IR $\nu^{\rm CC14}_{\rm max}$ cm⁻¹: CO₂R 1740; furan ketone 1687, 1565, 883. MS $\rm m_{\rm max}^{\rm max}$ (rel. int.): 224.105 (M⁺1) (calc. for C₁₂H₁₆O₄ 224.105); 193 (M⁺ \rightarrow OCH₃, 2). 150 (M⁺ \rightarrow

Eumorphia prostata Bolus (voucher 77/130): Roots (30 g) afford 2 mg 21 and 15 mg 2, while 80 g aerial parts yielded 50 mg 20, 1 mg 17 and 1 mg 18 (Et₂O-petrol, 1:10, AgNO₃ coated plate), 10 mg 5 (Et₂O-petrol 1:10), 6 mg 7, 15 mg 13 and 6 mg 16.

plate), 10 mg 5 (Et₂O-petrol 1:10), 6 mg 7, 15 mg 13 and 6 mg 16.

Eumorphistonol (16). Colourless oil, IR v^{CCl₄} cm⁻¹: OH 3620; furanketone 1685, 1570, 878. MS m/e (rel. int.): 233.118 (M⁺-CH₃, 1) (calc. for C₁₄H₁₇O₃ 233.118); 230 (M⁺-H₂O, 11);

10-Oxolasiospermane (5). Colourless oil, IR $v_{max}^{\text{CCI}_4}$ cm⁻¹: furan ketone 1680, 1563, 1510, 870. MS m/e (rel. int.): 246.126 (M⁺, 12) (calc. for $C_{15}H_{18}O_3$ 246.126); 217 (M⁺-'CHO, 2);

17. Colourless oil, IR $v_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: furan ketone 1685, 1570, 875. MS (GC-MS) m/e (rel. int.): 230 (M⁺, 20); 215 (M⁺ - 'CH₃,

18. Colourless oil, IR $\nu_{\text{max}}^{\text{CCl.4}}$ cm⁻¹: furan ketone 1685, 1570, 875. MS (GC-MS) m/e (rel. int.): 330 (M⁺, 13); 215 (M⁺ - 'CH₃, 17);

135 (
$$M^+$$
- \bigcirc \bigcirc \bigcirc CO $^+$, 15); 95 (\bigcirc \bigcirc CO $^+$, 100).

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